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# Macrocalyxin I

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The title compound, 2-[1,2,3,4,4a,4b,5,6,7,8,8a,9-dodecahydro-7-hydroxy-4b,8,8-trimethylphenanthren-2-yl]propenoic acid,  $C_{20}H_{30}O_3$ , is a naturally occurring diterpenoid which was isolated from *Rabdosia macrocalyx*. The hydroxy and carboxy groups, which are located at the two ends of the molecule, both serve as simultaneous hydrogen-bond donors and acceptors. Two intermolecular  $O-H\cdots O$  hydrogen bonds are present and link each molecule to four neighbours, thus forming an extensive hydrogen-bond network within the crystal.

# Comment

*Rabdosia macrocalyx* is widely distributed in Anhui, Jiangsu, Zhejiang, Jiangxi, Fujian, Hunan, Guangdong, Guangxi and Taiwan provinces, China, where it has been used as a folk medicine. Its decoctions are used as antibiotics and for antitumor treatment. Macrocalyxins A, B (Cheng *et al.*, 1984), C (Wang *et al.*, 1984), D (Wang *et al.*, 1985), E (Wang *et al.*, 1986), F, G and H (Wang *et al.*, 1995) have been isolated previously from this plant. In order to isolate more bioactive constituents, we investigated the whole herb of *Rabdosia macrocalyx*, which led to the isolation of the title compound, the natural diterpenoid macrocalyxin I, (1), which was isolated from *Rabdosia macrocalyx* Hara for the first time. Its structure was established from spectral evidence and was confirmed by the present X-ray diffraction study.



The molecule of (1) (Fig. 1) is composed of three sixmembered rings. Rings A (C1–C5/C10) and C (C8/C9/C12– C14) adopt a chair conformation, with mean torsion angles of 51.4 and 52.4°, respectively. Ring B (C5–C9/C10) adopts a





View of the title molecule showing the atomic numbering scheme and 50% probability displacement ellipsoids.





The intermoleculer hydrogen bonding in (1) viewed normal to the (001) plane. H atoms have been omitted for clarity, except for those involved in hydrogen bonds, which are shown as small spheres of arbitrary radii. Hydrogen bonds are shown as dashed lines.

half-chair conformation owing to the double bond between C7 and C8. The stereochemistry of the A/B ring junction is *trans*, and the dihedral angle between rings A and B is 17.60 (5)°; the



### Figure 3

The crystal-packing diagram for the title compound viewed down the c axis. H atoms have been omitted for clarity, except for these involved in hydrogen bonds.

dihedral angle between rings B and C is  $14.92(7)^{\circ}$ . The configurations at the other chiral centers are as follows: C3-OH, C10-Me, C9-H and C13-H are axial and the 1-carboxyethenyl group at C13 is equatorial.

The C17-C15-C16-O2 torsion angle has a value of 172.4 (3)° because of the conjugated double bound. The best least-squares plane formed by atoms C17/C15/C16/O2/O3 has a maximum deviation of 0.0592 Å, and the dihedral angle between this plane and ring C is  $114.22 (11)^{\circ}$ . The hydroxy group located at C3 and the carboxy group located at C15 participate in hydrogen bonding. Both groups serve as simultaneous hydrogen-bond donors and acceptors. Two intermolecular  $O-H \cdots O$  hydrogen bonds (Table 2) are present and link each molecule to four adjacent neighbours (Fig. 2). The overall result is an extended hydrogen-bonding network throughout the structure (Fig. 3).

# **Experimental**

Dried powder (7.5 kg) of the whole herb of Rabdosia macrocalyx was soaked three times with 95% EtOH at room temperature. The solvent was removed by evaporation at reduced pressure, and the residue was successively fractioned with petroleum ether, EtOAc and *n*-BuOH. The residue of the EtOAc fraction was subjected to column chromatography over silica gel. The column was eluted with a petroleum ether-EtOAc mixture. The crude compound was purified by column chromatography on silica gel with an acetone-chloroform mixture, producing 210 mg of macrocalyxin A and 60 mg of the pure title compound, (1) [m.p.: 490.5-492.5 K (CHCl<sub>3</sub>/CH<sub>3</sub>COCH<sub>3</sub>)]. <sup>13</sup>C NMR (125 MHz, pyridine): δ (p.p.m.) 169.8 (C16), 147.5 (C15), 137.2 (C8), 121.7 (C17), 121.2 (C7), 75.1 (C3), 52.4 (C9), 44.3 (C5), 41.7 (C6), 39.7 (C13), 37.5 (C4), 35.3 (C10), 32.3 (C14), 32.2 (C12), 29.3 (C19), 26.4 (C11), 25.7 (C1), 23.3 (C2), 23.1 (C18), 15.3 (C20). Crystals suitable for X-ray structure analysis were obtained by slow evaporation from an aqueous solution in chloroform and methanol (1:1) at room temperature.

### Crystal data

$C_{20}H_{30}O_3$	$D_x = 1.176 \text{ Mg m}^{-3}$
$M_r = 318.44$	Mo $K\alpha$ radiation
Monoclinic, C2	Cell parameters from 25
a = 24.066 (2) Å	reflections
b = 10.017 (1) Å	$\theta = 3.3 - 12.3^{\circ}$
c = 7.608 (1)  Å	$\mu = 0.08 \text{ mm}^{-1}$
$\beta = 101.35 \ (1)^{\circ}$	T = 295 (2)  K
V = 1798.2 (3) Å <sup>3</sup>	Prism, colorless
Z = 4	$0.50\times0.50\times0.40$ mm
Data collection	
Siemens P4 diffractometer	$h = 0 \rightarrow 31$
$\omega$ scans	$k = 0 \rightarrow 13$
2311 measured reflections	$l = -9 \rightarrow 9$
2178 independent reflections	3 standard reflections
1605 reflections with $I > 2\sigma(I)$	every 97 reflections
D 0.016	intensity deserve 1.20/

 $R_{int} = 0.016$  $\theta_{\rm max} = 27.5^{\circ}$ 

#### Refinement

Refinement on  $F^2$  $R[F^2 > 2\sigma(F^2)] = 0.040$  $wR(F^2) = 0.097$ S = 0.982178 reflections 214 parameters H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0509P)^2]$ where  $P = (F_o^2 + 2F_c^2)/3$  $(\Delta/\sigma)_{\rm max} < 0.001$  $\Delta \rho_{\rm max} = 0.15 \text{ e } \text{\AA}^{-3}$  $\Delta \rho_{\rm min} = -0.12 \text{ e} \text{ Å}^{-3}$ Extinction correction: SHELXL97 Extinction coefficient: 0.035 (2)

# Table 1

Selected geometric parameters (Å, °).

O1-C3	1.432 (3)	C13-C15	1.512 (3)
O2-C16	1.217 (3)	C15-C17	1.310 (4)
O3-C16	1.320 (3)	C15-C16	1.479 (3)
C7-C8	1.320 (3)		
O1-C3-C2	109.2 (2)	O2-C16-O3	121.8 (2)
O1-C3-C4	108.4 (2)	O2-C16-C15	123.2 (2)
C17-C15-C16	119.4 (2)	O3-C16-C15	115.0 (2)
C17-C15-C13	124.9 (2)		
C17-C15-C16-O2	172.4 (3)	C17-C15-C16-O3	-7.8 (4)

Та	ble	2		
* *				

Hydrogen-bonding geometry (Å, °).

$D - H \cdots A$	$D-{\rm H}$	$H \cdots A$	$D \cdot \cdot \cdot A$	$D - H \cdots A$
$O3-H3O\cdots O1^{i}$ $O1-H1O\cdots O2^{ii}$	0.82 0.82	1.85 1.97	2.662 (3) 2.771 (3)	169 164
$O1-H1O\cdots O2^n$	0.82	1.97	2.771 (3)	164

Symmetry codes: (i)  $x - \frac{1}{2}, \frac{1}{2} + y, z$ ; (ii)  $\frac{3}{2} - x, y - \frac{1}{2}, 2 - z$ .

H atoms were placed in geometrically calculated positions and included in the final refinement as riding, with  $U_{\rm iso}$  values equal to  $1.2U_{\rm eq}$  of the carrier atom. An attempt to establish the absolute configuration failed. The Flack (1983) parameter obtained was -0.9 (16). The Friedel pairs were merged before the final refinement and only the relative stereochemistry is shown in the Scheme and figures.

Data collection: XSCANS (Siemens, 1994); cell refinement: XSCANS; data reduction: XSCANS; program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: XP in SHELXTL/PC (Siemens, 1991); software used to prepare material for publication: SHELXTL/PC.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: TA1347). Services for accessing these data are described at the back of the journal.

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